



Mechanical	<i>Force acts upon an object</i>
Electrical	<i>Electric current flow</i>
Heat	<i>Temperature difference between objects</i>
Radiation	<i>Electromagnetic waves or sound</i>

Energy pathways

Change in thermal energy = mass X specific heat capacity X temperature change $\Delta E = m \times c \times \Delta \theta$

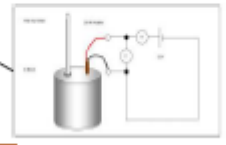
Specific Heat Capacity	<i>Energy needed to raise 1kg of substance by 1°C</i>	Depends on: mass of substance, what the substance is and energy put into the system.
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HIGHER: efficiency can be increased using machines.

Efficiency = $\frac{\text{Useful power output}}{\text{Total power input}}$

Efficiency = $\frac{\text{Useful output energy transfer}}{\text{Total input energy transfer}}$

Efficiency *How much energy is usefully transferred*



Kinetic energy	<i>Energy stored by a moving object</i>	$\frac{1}{2} \times \text{mass} \times (\text{speed})^2$ $\frac{1}{2} mv^2$
Elastic Potential energy	<i>Energy stored in a stretched spring, elastic band</i>	$\frac{1}{2} \times \text{spring constant} \times (\text{extension})^2$ $\frac{1}{2} ke^2$ (Assuming the limit of proportionality has not been exceeded)
Gravitational Potential energy	<i>Energy gained by an object raised above the ground</i>	Mass X gravitational field strength X height mgh

Energy stores and changes

AQA ENERGY – part 1

Energy Conservation and Dissipation

System	<i>An object or group of objects that interact together</i>	EG: Kettle boiling water.
Energy stores	<i>Kinetic, chemical, internal (thermal), gravitational potential, elastic potential, magnetic, electrostatic, nuclear</i>	Energy is gained or lost from the object or device.
Ways to transfer energy	<i>Light, sound, electricity, thermal, kinetic are ways to transfer from one store to another store of energy.</i>	EG: electrical energy transfers chemical energy into thermal energy to heat water up.
Unit	<i>Joules (J)</i>	

Dissipate *To scatter in all directions or to use wastefully*
When energy is 'wasted', it dissipates into the surroundings as internal (thermal) energy.



Ways to reduce 'wasted' energy
Energy transferred usefully
Insulation, streamline design, lubrication of moving parts.

Closed system	<i>No change in total energy in system</i>
Open system	<i>Energy can dissipate</i>

Principle of conservation of energy
The amount of energy always stays the same.
Energy cannot be created or destroyed, only changed from one store to another.

Work	<i>Doing work transfers energy from one store to another</i>	By applying a force to move an object the energy store is changed.	Work done = Force X distance moved $W = Fs$
Power	<i>The rate of energy transfer</i>	1 Joule of energy per second = 1 watt of power	Power = energy transfer ÷ time $P = E \div t$ Power = work done ÷ time, $P = W \div t$



HIGHER: When an object is moved, energy is transferred by doing work.

Work done = Force X distance moved

Frictional forces cause energy to be transferred as thermal energy. This is wasted.

Reducing friction - using wheels, applying lubrication. Reducing air resistance - travelling slowly, streamlining.

	<i>Units</i>
Specific Heat Capacity	<i>Joules per Kilogram degree Celsius (J/Kg°C)</i>
Temperature change	<i>Degrees Celsius (°C)</i>
Work done	<i>Joules (J)</i>
Force	<i>Newton (N)</i>
Distance moved	<i>Metre (m)</i>
Power	<i>Watts (W)</i>
Time	<i>Seconds (s)</i>

Useful energy	<i>Energy transferred and used</i>	
Wasted energy	<i>Dissipated energy, stored less usefully</i>	

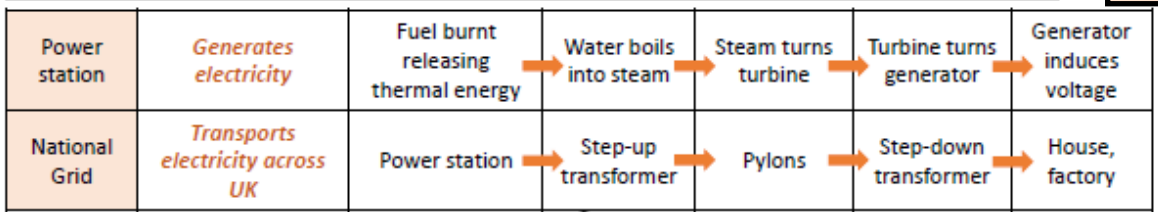
Prefix	<i>Multiple</i>	Standard form
Kilo	<i>1000</i>	10^3
Mega	<i>1000 000</i>	10^6
Giga	<i>100 000 000</i>	10^9

	<i>Units</i>
Energy (KE, EPE, GPE, thermal)	<i>Joules (J)</i>
Velocity	<i>Metres per second (m/s)</i>
Spring constant	<i>Newton per metre (N/m)</i>
Extension	<i>Metres (m)</i>
Mass	<i>Kilogram (Kg)</i>
Gravitational field strength	<i>Newton per kilogram (N/Kg)</i>
Height	<i>Metres (m)</i>

Using renewable energy will need to increase to meet demand.

Transport	<i>Petrol, diesel, kerosene produced from oil</i>	Used in cars, trains and planes.
Heating	<i>Gas and electricity</i>	Used in buildings.
Electricity	<i>Most generated by fossil fuels</i>	Used to power most devices.

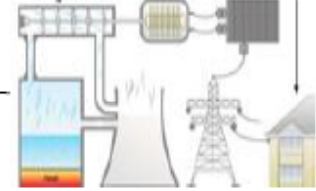
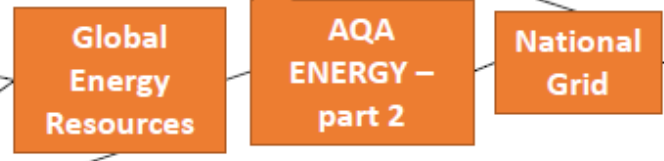
Power station – NB: You need to understand the principle behind generating electricity. An energy resource is burnt to make steam to drive a turbine which drives the generator.



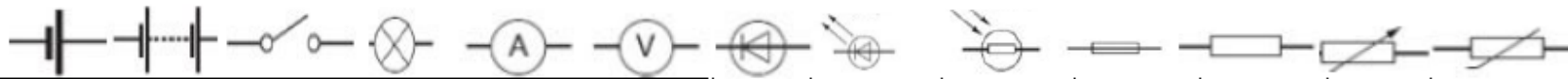
Renewable energy makes up about 20% of energy consumption. Fossil fuel reserves are running out. Energy demand is increasing as population increases.

Non-renewable energy resource	<i>These will run out. It is a finite reserve. It cannot be replenished.</i>	e.g. Fossil fuels (coal, oil and gas) and nuclear fuels.
Renewable energy resource	<i>These will never run out. It is an infinite reserve. It can be replenished.</i>	e.g. Solar, Tides, Waves, Wind, Geothermal, Biomass, Hydroelectric

Using fuels
Energy resources

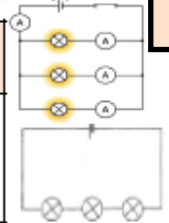


Energy resource	How it works	Uses	Positive	Negative
Fossil Fuels (coal, oil and gas)	<i>Burnt to release thermal energy used to turn water into steam to turn turbines</i>	Generating electricity, heating and transport	Provides most of the UK energy. Large reserves. Cheap to extract. Used in transport, heating and making electricity. Easy to transport.	Non-renewable. Burning coal and oil releases sulfur dioxide. When mixed with rain makes acid rain. Acid rain damages building and kills plants. Burning fossil fuels releases carbon dioxide which contributes to global warming. Serious environmental damage if oil spilt.
Nuclear	<i>Nuclear fission process</i>	Generating electricity	No greenhouse gases produced. Lots of energy produced from small amounts of fuel.	Non-renewable. Dangers of radioactive materials being released into air or water. Nuclear sites need high levels of security. Start up costs and decommission costs very expensive. Toxic waste needs careful storing.
Biofuel	<i>Plant matter burnt to release thermal energy</i>	Transport and generating electricity	Renewable. As plants grow, they remove carbon dioxide. They are 'carbon neutral'.	Large areas of land needed to grow fuel crops. Habitats destroyed and food not grown. Emits carbon dioxide when burnt thus adding to greenhouse gases and global warming.
Tides	<i>Every day tides rise and fall, so generation of electricity can be predicted</i>	Generating electricity	Renewable. Predictable due to consistency of tides. No greenhouse gases produced.	Expensive to set up. A dam like structure is built across an estuary, altering habitats and causing problems for ships and boats.
Waves	<i>Up and down motion turns turbines</i>	Generating electricity	Renewable. No waste products.	Can be unreliable depends on wave output as large waves can stop the pistons working.
Hydroelectric	<i>Falling water spins a turbine</i>	Generating electricity	Renewable. No waste products.	Habitats destroyed when dam is built.
Wind	<i>Movement causes turbine to spin which turns a generator</i>	Generating electricity	Renewable. No waste products.	Unreliable – wind varies. Visual and noise pollution. Dangerous to migrating birds.
Solar	<i>Directly heats objects in solar panels or sunlight captured in photovoltaic cells</i>	Generating electricity and some heating	Renewable. No waste products.	Making and installing solar panels expensive. Unreliable due to light intensity.
Geothermal	<i>Hot rocks under the ground heats water to produce steam to turn turbine</i>	Generating electricity and heating	Renewable. Clean. No greenhouse gases produced.	Limited to a small number of countries. Geothermal power stations can cause earthquake tremors.



Electrons carry current. Electrons are free to move in metal.

Cell	Battery	Switch	Lamp	Ammeter	Volt meter	Diode	LED	LDR	Fuse	Resistor	Variable resistor	Thermistor
Store of chemical energy	Two or more cells in series	Breaks circuit, turning current off	Lights when current flows	Measures current	Measures potential difference	Current flows one way	Emits light when current flows	Resistance low in bright light	Melts when current is too high	Affects the size of current flowing	Allows current to be varied	Resistance low at high temp



Current	Flow of electrical charge	Ampere (A)
Potential difference (p.d.)	How much electrical work is done by a cell	Volts (V)
Charge	Amount of electricity travelling in a circuit	Coulombs (C)

Circuit symbols

Current and Charge
Current, potential difference and resistance

Series and parallel circuits

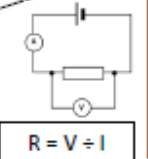
Series circuit	Current is the same in all components.	Total p.d. from battery is shared between all the components.	Total resistance is the sum of each component's resistance.
Parallel circuit	Total current is the sum of each component's current.	p.d. across all components is the same.	Total resistance is less than the resistance value of the smallest individual resistor.

Series	Parallel
A circuit with one loop	A circuit with two or more loops
Total p.d. If cells are joined in series, add up individual cell values	

Charge = Current X time $Q = I \times t$

Changing current
 Change the p.d. of the cells
 Add more components

Controlling current



$R = V \div I$
 Resistance = Potential difference \div Current

Ammeter	Set up in series with components
Voltmeter	Set up parallel to components

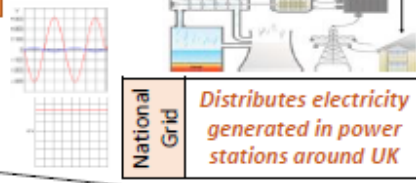
Resistance (Ω)	A measurement of how much current flow is reduced
The higher the resistance, the more difficult it is for current to flow.	
Increasing resistance, reduces current.	
Increasing voltage, increases current.	

AQA Electricity

Energy transfers

Power (W) = potential difference X current $R = V \times I$
 Work is done when charge flowing.
 Power = (current)² X resistance $P = I^2 \times R$
 Energy transferred = Power X time $E = P \times t$

Domestic uses and safety



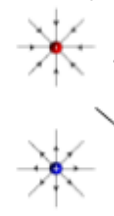
Step-up transformers	Step-down transformers
Increase voltage, decrease current	Decrease voltage, increase current
Increases efficiency, reduces heat loss.	Makes safer for houses.

Ohmic conductor	At a constant temperature, current is directly proportional to the p.d. across the resistor.
Filament lamp	As current increases, the resistance increases. The temperature increases as current flows.
Diode	Current flows when p.d. flows forward. Very high resistance in reverse.

Current - Potential difference graphs

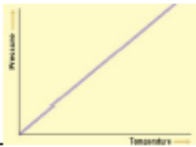
3 pin plug	Live - Brown	Carries p.d from mains supply.	p.d. between live and earth = 230V
	Neutral - Blue	Completes the circuit.	p.d. = 0V
	Earth - Green and Yellow stripes	Only carries current if there is a fault.	p.d. = 0V

Like charges	Repel
Unlike charges	Attract



Static electricity - PHYSICS only

Static electricity	Electrical charge is stationary	When two insulating material are rubbed together, electrons move from one material to the other.
Shocks	Walking on carpet causes friction. Electrons move to the person and charge builds up. When the person touches a metal object, the electrons conduct away, making a spark.	
Electric fields	Charged objects create electric fields around them. Strongest closest to the object. The field direction is the direction of force on a positive charge. Add more charge increases field strength.	



Pressure of a fixed volume of gas increases as temperature increases (temperature increases, speed increases, collisions occur more frequently and with more force so pressure increases).

Temperature of gas is linked to the average kinetic energy of the particles.

If kinetic energy increases so does the temperature of gas.

No kinetic energy is lost when gas particles collide with each other or the container.

Gas particles are in a constant state of random motion.

$$P = m \div V$$

Density = mass \div volume.

Density *Mass of a substance in a given volume*

Kinetic theory of gases

State	Particle arrangement	Properties
Solid	<i>Packed in a regular structure. Strong forces hold in place so cannot move.</i>	Difficult to change shape.
Liquid	<i>Close together, forces keep contact but can move about.</i>	Can change shape but difficult to compress.
Gas	<i>Separated by large distances. Weak forces so constantly randomly moving.</i>	Can expand to fill a space, easy to compress.

	Units
Density	<i>Kilograms per metre cubed (kg/m³)</i>
Mass	<i>Kilograms (kg)</i>
Volume	<i>Metres cubed (m³)</i>
Energy needed	<i>Joules (J)</i>
Specific latent heat	<i>Joule per kilogram (J/kg)</i>
Change in thermal energy	<i>Joules (J)</i>
Specific heat capacity	<i>Joule per kilogram degrees Celsius (J/kg°C)</i>
Temperature change	<i>Degrees Celsius (°C)</i>
Pressure	<i>Pascals (Pa)</i>

P3

Particle model

Pressure

AQA PARTICLE MODEL OF MATTER

PHYSICS ONLY: when you do work the temperature increases e.g. pump air quickly into a ball, the air gets hot because as the piston in the pump moves the particles bounce off increasing kinetic energy, which causes a temperature rise.

Reducing the volume of a fixed mass of gas increases the pressure.

Halving the volume doubles the pressure.

PV = constant.

$$P_1V_1 = P_2V_2$$

Internal energy and energy transfers

Specific Heat Capacity *Energy needed to raise 1kg of substance by 1°C*

Depends on:

- Mass of substance
- What the substance is
- Energy put into the system.

Change in thermal energy = mass X specific heat capacity X temperature change.

$$\Delta E = m \times c \times \Delta \theta$$

Change of state

Specific Latent Heat	<i>Energy needed to change 1kg of a substance's state</i>
Specific Latent Heat of Fusion	<i>Energy needed to change 1kg of solid into 1 kg of liquid at the same temperature</i>
Specific Latent Heat of Vaporisation	<i>Energy needed to change 1kg of liquid into 1 kg of gas at the same temperature</i>

Internal energy

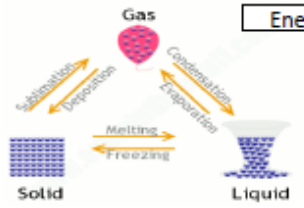
Energy stored inside a system by particles

Internal energy is the total kinetic and potential energy of all the particles (atoms and molecules) in a system.

Heating changes the energy stored within a system

Heating causes a change in state. As particles separate, potential energy stored increases. Heating increases the temperature of a system. Particles move faster so kinetic energy of particles increases.

Freezing	Liquid turns to a solid. Internal energy decreases.
Melting	Solid turns to a liquid. Internal energy increases.
Boiling / Evaporating	Liquid turns to a gas. Internal energy increases.
Condensation	Gas turns to a liquid. Internal energy decreases.
Sublimation	Solid turns directly into a gas. Internal energy increases.
Conservation of mass	When substances change state, mass is conserved.
Physical change	No new substance is made, process can be reversed.



Energy needed = mass X specific latent heat.

$$\Delta E = m \times L$$



Radius of an atom
 $1 \times 10^{-10} \text{m}$



Electrons gained
Negative ion

Electrons lost
Positive ion

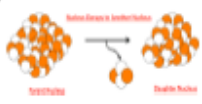
Atom	Same number of protons and electrons
Ion	Unequal number of electrons to protons
Mass number	Number of protons <u>and</u> neutrons
Atomic number	Number of protons

Particle	Charge	Size	Found
Neutron	None	1	In the nucleus
Proton	+	1	
Electron	-	Tiny	

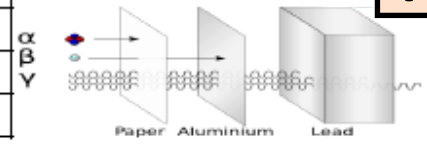
Isotope	${}^6_3\text{Li}$		${}^7_3\text{Li}$	
Different forms of an element with the same number of protons but different number of neutrons				

Discovery of the nucleus

Democritus	Suggested idea of atoms as small spheres that cannot be cut.
J J Thomson (1897)	Discovered electrons – emitted from surface of hot metal. Showed electrons are negatively charged and that they are much less massive than atoms.
Thomson (1904)	Proposed 'plum pudding' model – atoms are a ball of positive charge with negative electrons embedded in it.
Geiger and Marsden (1909)	Directed beam of alpha particles (He^{2+}) at a thin sheet of gold foil. Found some travelled through, some were deflected, some bounced back.
Rutherford (1911)	Used above evidence to suggest alpha particles deflected due to electrostatic interaction between the very small charged nucleus, nucleus was massive. Proposed mass and positive charge contained in nucleus while electrons found outside the nucleus which cancel the positive charge exactly.
Bohr (1913)	Suggested modern model of atom – electrons in circular orbits around nucleus, electrons can change orbits by emitting or absorbing electromagnetic radiation. His research led to the idea of some particles within the nucleus having positive charge; these were named protons.
Chadwick (1932)	Discovered neutrons in nucleus – enabling other scientists to account for mass of atom.

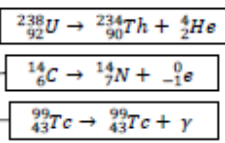


Decay	Range in air	Ionising power	Penetration power
Alpha	Few cm	Very strong	Stopped by paper
Beta	Few m	Medium	Stopped by Aluminium
Gamma	Great distances	Weak	Stopped by thick lead



Radioactive decay	Unstable atoms randomly emit radiation to become stable
Detecting	Use Geiger Muller tube
Unit	Becquerel
Ionisation	All radiation ionises

Decay	Emitted from nucleus	Changes in mass number and atomic number	
Alpha (α)	Helium nuclei (${}^4_2\text{He}$)	-4	-2
Beta (β)	Electron (${}^0_{-1}\text{e}$)	0	+1
Gamma (γ)	Electromagnetic wave	0	0
Neutron	Neutron	-1	0



Atoms and Isotopes

Atoms and Nuclear Radiation

Contamination	Unwanted presence of radioactive atoms
Irradiation	Person is in exposed to radioactive source

AQA ATOMIC STRUCTURE

PHYSICS ONLY: Hazards and uses of Radioactive emissions and of background radiation

Half life	The time taken to lose half of its initial radioactivity
Sievert	Unit measuring dose of radiation
Background	Constant low level environmental radiation, e.g. from nuclear testing, nuclear power, waste

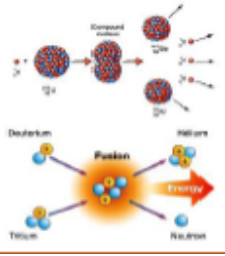
Nuclear fission and fusion

Uses	Different isotopes have different half lives	Short half-lives used in high doses, long half lives used in low doses.
Tracers	Used within body	Isotope with short half life injected, allowed to circulate and collect in damaged areas. PET scanner used to detect emitting radiation. Must be beta or gamma as alpha does not penetrate the body.
Radiation therapy	Used to treat illnesses e.g. cancer	Cancer cells killed by gamma rays. High dose used to kill cells. Damage to healthy cells prevented by focussed gamma ray gun.

Fuel rods	Made of U-238, 'enriched' with U-235 (3%). Long and thin to allow neutrons to escape, hitting nuclei.
Control rods	Made of Boron. Controls the rate of reaction. Boron absorbs excess neutrons.
Concrete	Neutrons hazardous to humans – thick concrete shield protects workers.

PHYSICS ONLY: Nuclear energy

Nuclear fission	One large unstable nucleus splits to make two smaller nuclei	Neutron hits U-235 nucleus, nucleus absorbs neutron, splits emitting two or three neutrons and two smaller nuclei. Process also releases energy.	Process repeats, chain reaction formed
Nuclear fusion	Two small nuclei join to make one larger nucleus	Difficult to do on Earth – huge amounts of pressure and temperature needed.	Occurs in stars





Each Kg has a gravitational pull of 9.8N.

Gravitational field strength	Gravity exerted around an object.	Earth's gfs = 9.8N/kg
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Unit	Newton (N)	1N
Kilo	Kilonewton (KN) = 1000	1X 10 ³
Mega	Meganewton (MN) = 1000,000	1 X 10 ⁶

Centre of mass
The weight of an object acts through a single point

Force	Push or pull	Stretch, squash, turn.
Contact force	Exerted between two objects when they touch	Friction, air resistance, tension.
Non-contact force	Exerted between two objects without touching	Gravity, electrostatic forces, magnetic forces.

Resolving forces
An object pulled with a force at an angle
A single force can be split into two components acting at right angles to each other.

The component forces combined have the same effect.

Weight = mass X gravitational field strength
 $W = m \times g$

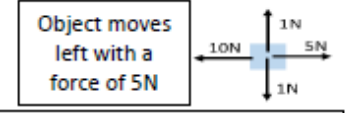
Weight	Force acting upon an object due to gravity	Newton (N)
Mass	How much matter	Kilograms (Kg)

Gravity

Resultant force
The overall effect of all of the forces acting upon an object
Two forces acting in the same direction are added.
Two forces acting in the opposite direction are taken away.

HIGHER ONLY
Work done against frictional forces, temperature of object rises.

Free body diagram
Show magnitude and direction of all forces upon an object



Forces and their interactions

AQA FORCES – part 1

PHYSICS ONLY

Scalar	A quantity that only has magnitude (size)	e.g. mass, time, speed, temperature, energy,
Vector	A quantity that only has magnitude and direction	e.g. force, velocity, momentum

An arrow can be used to show vectors
Length of arrow = magnitude of vector
Direction of arrow = direction of vector



Scalar and vector quantities
Moments, levers and gears

$M = F \times d$
Moment = force X distance

Velocity	Speed + direction	The speed of a car is 30m/s. A car moves forward with a velocity of 30m/s
Distance	How far	The table is 1m long
Displacement	Distance + direction	The beach is 1km due east of the town

Moment
Turning effect of a force about a pivot
Lever
A small force exerted with a long lever applies a large force

Area	Metres squares (m ²)
Weight	Newton (N)
Mass	Kilograms (kg)
Gravitational field strength	Newton per kilogram (N/Kg)
Force	Newton (N)
Work done	Joules (J)
Distance	Metres (m)
Moment	Newton-metres (Nm)

Gears
Increase or decrease the rotational effect of a force

Principle of moments
In a balanced system, the sum of the clockwise moments = the sum of the anti-clockwise moments

Pressure = height X density X gfs

HIGHER ONLY
Pressure
Pressure = Force ÷ Area
 $P = F \div A$

Pressure and depth
Pressure on divers depends on weight of water above

Upthrust
Resultant force exerted by a fluid

Fluid
A liquid or gas
Flows and changes shape to fill a container.

Hydraulic machine
Use liquids to transmit pressure

Work done and energy transfer
Work done = force X distance moved
 $W = F \times s$
1J of work is done when 1N of force moves an object through a distance of 1m, in the direction of the force.

If force is at right angles to direction of movement, NO work is done.

Forces and elasticity
One force
The object changes speed or direction
More than one force
The object changes shape

Two balanced forces can stretch a object.
Two balanced forces can compress an object.
Three balanced forces can bend an object.

Elastic deformation	The object has been stretched but returns to its original length	Limit of proportionality Beyond this point the spring is permanently deformed
Inelastic deformation	The object has been stretched but does not return to its original length	
Extension	The difference between stretched and unstretched lengths	

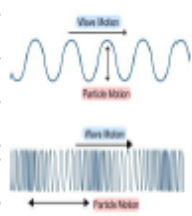
Stretching a spring
Force = spring constant X extension, $F = k \times e$
 $EPE = \frac{1}{2} \times \text{spring constant} \times (\text{extension})^2$, $EPE = \frac{1}{2} ke^2$

Elastic Potential energy (EPE) Energy stored in a stretched spring

Force	Newton (N)
Spring constant	Newton per metre (N/m)
Extension	Metres (m)
EPE	Joules (J)

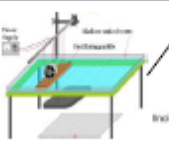
Atmospheric pressure
Caused by billions of air particles colliding with a surface.

Wave speed	Wave speed = frequency X wavelength	$V = f \times \lambda$
Wave period	Wave period = $1 \div$ frequency	$T = 1 \div f$
Speed	Speed = distance \div time	$v = d \div t$



Transverse wave	Vibration causing the wave is at right angles to the direction of energy transfer	Energy is carried outwards by the wave.	Water and light waves, S waves.
Longitudinal wave	Vibration causing the wave is parallel to the direction of energy transfer	Energy is carried along the wave.	Sound waves, P waves.

Wavelength	Distance from one point on a wave to the same point of the next wave
Amplitude	The maximum disturbance from its rest position
Frequency	Number of waves per second
Period	Time taken to produce 1 complete wave



Measuring speed

- In water, use a ripple tank.
- In air, use echoes.

Properties

Air Water

Sound waves travelling through different mediums, the frequency stay constant.

Transverse and Longitudinal waves

Waves in air, fluids and solids

AQA Waves

Black body radiation

PHYSICS ONLY

Earth and Global warming

Ultraviolet, visible light, infra-red radiation penetrate atmosphere and heat up Earth's surface.

Longer wavelengths are radiated back, trapped by atmosphere.

Energy lost is not at the same rate as energy being absorbed so Earth heats up.

Black body radiation

All objects absorb or reflect infrared radiation

Hotter objects emit more infrared radiation.

Constant temperature

Rate of absorption = rate of radiation

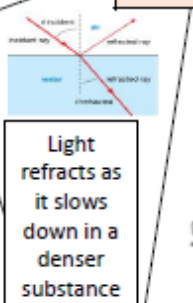
Intensity and wavelength of energy affects temperature.

e.g. Gamma

Short wavelengths have high frequency and high energy.

Angle of incidence = angle of reflection (i) = (r)

Reflection	Wave bounces off the surface.
Refraction	Waves changes direction at boundary.
Transmitted	Passes through the object.
Absorbed	Passes into but not out of, transfers energy and heats up the object.



Electromagnetic waves

Continuous spectrum of transverse waves

gamma ray ultraviolet X-ray visible infrared microwave radio

Magnification = image size \div object size

PHYSICS HIGHER ONLY

Hearing

Frequencies between 20 - 20,000 Hz

Longitudinal waves cause ear drum to vibrate, amplified by three ossicles which creates pressure in the cochlea.

Absorbed light changes into thermal energy store.

PHYSICS ONLY

	Units
Distance	Metres (m)
Wave speed	Metres per second (m/s)
Wavelength	Metres (m)
Frequency	Hertz (Hz)
Period	Seconds (s)

Seismic waves

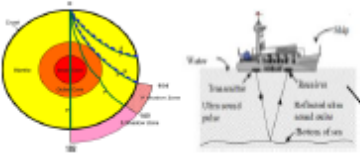
P wave	S wave	Seismograph
Longitudinal	Transverse	Shows P and S waves arriving at different times.
Fast	Slow	
Travel through solids and liquids	Travels through solids	By using the times the waves arrive at the monitoring centres, the epicentre of earthquake can be found. ($v = x \div t$).
Produced by earthquakes.		

Black surfaces	Good emitters, good absorbers
White surfaces	Poor emitters, poor absorbers
Shiny surfaces	Good reflectors

EM waves refract

HIGHER: Properties

Convex	Real or virtual images.	2F	Image same size, upside down, real.	Specular	Flat surface reflection.
Concave	Only virtual images.	2F - F	Image larger, upside down, real.		
		< F	Image bigger, right way, virtual.		
				Diffuse	Rough surface reflection.



Ultra sound	Partially reflected off boundary	Used for medical and foetal scans.
Sonar	Reflected off objects	Used to determine depth of objects under the sea.

EM wave	Danger	Use	Low frequency, long wavelength.
Radio	Safe.	Communications, TV, radio.	
Microwave	Burning if concentrated.	Mobile phones, cooking, satellites.	White Wave lengths reflected
Infrared		Heating, remote controls, cooking.	
Visible	Damage to eyes.	Illumination, photography, fibre optics.	Black Wave lengths absorbed
Ultra violet	Sunburn, cancer.	Security marking, disinfecting water.	
X-ray	Cell destruction, mutation, cancer.	Broken bones, airport security.	High frequency, short wavelength
Gamma		Sterilising, detecting and killing cancer.	



Relay
A device using a small current to control a larger current in another circuit
 Solenoid is wound around an iron core. Small current magnetises the solenoid. This attracts to electrical contacts, making a complete circuit. Current flows from battery to starter motor.

Split-ring commutator
Split ring touching two carbon brush contacts

Loud speakers
Converts variations in electrical current into sound waves.

Varying current flows through a coil that is in a magnetic field. A force on the wire moves backwards and forwards as current varies. Coil connected to a diaphragm. Diaphragm movements produce sound waves.



Electromagnet
Lots of turns of wire increase the magnetising effect when current flows
 Turn current off, magnetism lost.

Increase strength of magnetic field
Use larger current
Use more turns of wire
Put turns of wire closer together
Use iron core in middle

Generators
Coil of wire rotating inside a magnetic field. The end of the coil is connected to slip rings.

Produces altering current.

Microphones
Converts pressure variations in sound waves into variations in current in electrical circuits.

Fleming's left-hand rule
 To predict the direction a straight conductor moves in a magnetic field.

Thumb	Direction of movement.
First finger	Direction of magnetic field.
Second finger	Direction of current.



Solenoid
A long coil of wire
 Magnetic field from each loop adds to the next.

Right hand rule
 Thumb: Direction of current.
 Fingers: Direction of magnetic field.

Magnetic field around a wire

AQA MAGNETISM AND ELECTROMAGNETISM

Induced potential, transformers and National Grid

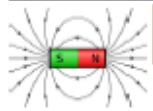
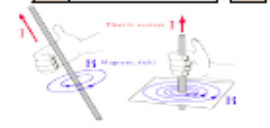
Reverse current, magnetic field direction reverses.

Further away from the wire, magnetic field is weaker.

Current large enough, iron filings show circular magnetic field.

If current is small, magnetic field is very weak.

Electric current flowing in a wire produces a magnetic field around it.



Permanent and Induced Magnetism

Magnets

Magnetic	<i>Materials attracted by magnets</i>	Uses non-contact force to attract magnetic materials.
North seeking pole	<i>End of magnet pointing north</i>	Compass needle is a bar magnet and points north.
South seeking pole	<i>End of magnet pointing south</i>	Like poles (N – N) repel, unlike poles (N – S) attract.
Magnetic field	<i>Region of force around magnet</i>	Strong field, force big. Weak field, force small. Field is strongest at the poles.
Permanent	<i>A magnet that produces its own magnetic field</i>	Will repel or attract other magnets and magnetic materials.
Induced	<i>A temporary magnet</i>	Becomes magnet when placed in a magnetic field.

National Grid
Distributes electricity generated in power stations around UK

PHYSICS HIGHER only

Transformer
Two coils of wire onto an iron core
 Alternating current supplied to primary coil, making magnetic field change. Iron core becomes magnetised, carries changing magnetic field to secondary coil. This induces p.d.

Step-up transformers <i>Increase voltage, decrease current</i> Increases efficiency by reducing amount of heat lost from wires.	Step-down transformers <i>Decrease voltage, increase current</i> Makes safer value of voltage for houses and factories.
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Induced potential
When a conducting wire moves through a magnetic field, p.d. is produced

Generator effect
Generates electricity by inducing current or p.d.

Uses of the generator effect
Dynamo, Microphones

Power lost = Potential difference X Current

Power supplied to primary coil = power supplied to secondary coil
 $V_p \times I_p = V_s \times I_s$

Voltage across the coil X number of coils (primary) = Voltage across the coil X number of coils (secondary)
 $V_p \div V_s = n_p \div n_s$

Force
Newton (N)

Force	<i>Newton (N)</i>
Magnetic flux density	<i>Tesla (T)</i>
Current	<i>Amperes (A)</i>
Length	<i>Metres (m)</i>
Power	<i>Watts (W)</i>
p.d.	<i>Voltage (V)</i>

Motor effect
HIGHER only
 Magnetic fields from the permanent magnet and current in the foil interact. This is called the motor effect.

$F = B \times I \times l$
 Force = magnetic flux density X current X length

If current and magnetic field are parallel to each other, no force on wire.

Reverse the current, foil moves upwards.

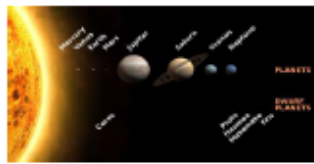
Aluminium foil placed between two poles of a strong magnet, will move downwards when current flows through the foil.

Size of force acting on foil depends on magnetic flux density between poles, size of current, length of foil between poles.

Magnetic flux
Lines drawn to show magnetic field
 Lots of lines = stronger magnets.

Magnetic flux density
Number of lines of magnetic flux in a given area
 Measures the strength of magnetic force.

Planet	<i>A large body orbiting the Sun</i>
Moon	<i>A natural satellite orbiting a planet</i>
Dwarf planet	<i>A body large enough to have its own gravity which caused a spherical shape</i>
Solar system	<i>Any object orbiting the Sun due to gravity</i>
Galaxy	<i>Collection of billions of stars</i>
Universe	<i>Collection of galaxies</i>



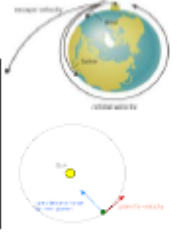
Comets, asteroids, satellites.
Other objects.

Effect of gravity.
Gravity causes moons to orbit planets, planets to orbit the Sun, stars to orbit galaxy centres.
Force of gravity changes the moon's direction not its speed.
Gravity pulls objects towards the ground.

Too fast = disappears into Space.
Correct speed = steady orbit around Earth.

Too slow = falls to Earth.

To calculate speed of Orbit: distance object moves in 1 orbit, Distance = $2\pi r$, then average speed = distance ÷ time.



Milky Way our galaxy.

Solar system

Orbital motions

Speed of Orbit.

HIGHER: Circular orbits.

Velocity = a vector.
A planet's velocity changes but speed remains constant.

Planets close to the Sun, gravity pull is strong. Planets move quickly.

Planets further away from the Sun, gravity pull is weaker. So speed of planet is slower.

Due to the Sun's gravity, planets accelerate towards the Sun and so changes direction.

When ambulances go past the sound changes from a high pitch to a low pitch.

Frequency of sound wave decreases, wavelength increases.

The life cycle of a star.

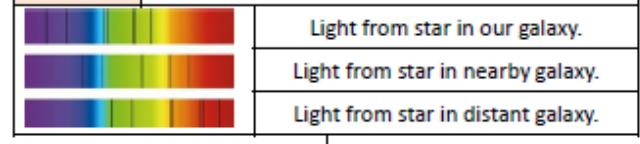
Nebula	<i>A cloud of cold hydrogen gas and dust</i>	Cloud collapses due to gravity, particles move very fast colliding with each other, kinetic energy transfers into internal energy and the temperature increases.
Protostar	<i>The large ball of gas contracts to form a star</i>	High temperature causes Hydrogen nuclei to collide and nuclear fusion begins. A star is 'born'.
Main sequence	<i>Stable period of star</i>	Gravity tries to collapse the star but enormous pressure of fusion energy expands and balances the inward force.

AQA SPACE PHYSICS PHYSICS ONLY

Red shift

Red-shift *The observed increase in wavelength of light from most distant galaxies. Light moves towards the red end of the spectrum.*

Hubble (1929) *He studied light from distant galaxies; found as frequency decreases, wavelength increases.*



Galaxies are moving away from us in all directions.

Light from distant galaxies is red-shifted, so galaxy is moving away from us.

Galaxies further away have bigger red-shift so are moving faster away.

Stars the same size as our Sun.

Red giant	<i>A large star that fuses Helium into heavier elements</i>	Hydrogen runs out, star becomes unstable, pressure inside drops causing star to collapse. Atoms now closer together results in atoms fusing and temperature increases. This increase in temperature causes the core to swell.
White dwarf	<i>Star collapses</i>	Nuclear fuel runs out, fusion stops, dense very hot core.
Black dwarf	<i>Cold dark star</i>	White dwarf cools down.

Stars larger than our Sun.

Red super giant	<i>Star swells greatly</i>	Nuclear fuel begins to run out and star swells (more matter = bigger size).
Supernova	<i>Gigantic explosion due to run away fusion reactions</i>	Rapid collapse, heats to very high temperatures causing run away nuclear reactions, star explodes, flinging remnants out into space. Large gravitational forces collapse the core into a tiny space. Remains of supernova form heavier elements (Iron and above)
Neutron star	<i>Very dense star</i>	Made out of neutrons.

The Big Bang *Universe began 13.8 billion years ago*
All matter and space expanded violently from a single point. Red—shift provides evidence for expansion.

Aristotle (ancient Greek)	<i>Earth at the centre, other heavenly bodies move around the Earth.</i>
Copernicus (1473 - 1543)	<i>Sun at the centre, other heavenly bodies move around the Sun.</i>
Galileo (1610)	<i>Made a telescope, looked at Jupiter, found four moons rotating around planet.</i>

Planets and moons moved at different speeds to stars = reason for different positions.

OR if collapse is into a really tiny space. **Black hole** *No light escapes* Gravitational forces so strong everything is pulled in.